Comparative studies on role of synthetic and biologically derived surface-active agents on removal of oil from oil spilled soil

Suganthi B*, Bharathidasan R

INTRODUCTION

Soil contamination by oil-based goods has become one of the principal issues because the synthetics and perilous materials they contain are a wellspring of worry for the whole ecosystem, so it is vital to track down an answer to this issue [1]. Oil is one of the main wellsprings of energy in the cutting edge modern world. However long the oil is presently double-dealing, transport, stockpiling and use, there is a gamble of spillage. Oil slicks represent a serious natural issue [2]. On 28 January 2017, an oil slick happened at Kamarajar Port in Chennai, Tamil Nadu. Despite the fact that legislatures and NGOs have gone to essential lengths to safeguard human existence, the effect has not been totally wiped out. Various methods have been developed to remove oil from contaminated areas. Mechanical lifting over oil using sorbents is certain of the close pregnant counter measures. Oil spills the oceans are over a massive situation due in imitation of their poor environmental conditions, it is necessary to rapidly identify vulnerable areas [3].

Therefore, to the extent that a new coastal decision support system considers environmental conditions, it is necessary to rapidly identify vulnerable areas or resources. Oil spills and their impact on this environment will reach and affect specific aspects of interest within the study area. Thus, the study addresses this gap by adopting the following model and predicts an oil slick path in the study area. In the present study, it resulted in the formation of biosurfactants or herbal surfactants similar to bacteria as a result of microorganisms. Biosurfactants have similar properties to surfactants from marine sediments of Ennore harbor, Chennai, Tamil Nadu and these isolates were identified as Bacillus genera with significant differences in their morphology. For the maximum production of surface-active agents, nutritional optimization in 6 screened isolates revealed that sucrose and yeast extract were the suitable carbon and nitrogen sources for growth in pH of 7.2 ± 0.2, at 37°C temperature and agitation at 180-200 rpm. Extraction and characterization studies revealed, the product was polymeric in nature with surface activity (28 ± 4 mN/m), thermal and emulsification stability (more than 90 days) when compared with synthetic surfactants. Based on laboratory scale studies, 90% of crude oil was removed from the aqueous phase within 60-120 minutes of exposure to the partially purified surface-active agent. This study suggested that the Bacillus species can be utilized for cleaning up hydrocarbon contamination sites and currently produce bio surfactants.

Key Words: Bioremediation; Crude oil removal; Biosurfactant; Surface tension; Oil contaminated soil
Removal of oil in aqueous and soil

Optimization of cell and oil concentrations for the oil removal: Growth medium of 50 ml sterile Zobell broth in 500 ml flask having different oil concentrations 2, 20, 200 and 2000 µl of stock solution containing 8, 80, 800 and 8000 × 10^6 of cells respectively, various amounts of oil was added (0.5, 1, 2.5, 5, 7.5 and 10 ml) to different concentration of cell culture and it was incubated individually for 48 hours in an orbital shaker at 37°C. After 48 hours, the flask culture was centrifuged and the supernatant was subjected to oil removal analysis.

Preparation of oil contaminated soil

Different types of soil (sand, beach, clay and clay loamy) were collected. After sampling, the soil was completely air-dried and sieved on a 10-mesh (<2 mm) and was screened prior to use. The sieved soil sample was analyzed for particles size distribution and soil texture was sandy soil, beach soil, clay soil and clay loamy soil. The soil was autoclaved three times to avoid the other bacterial contamination. In a 1 L beaker, 100 g of air-dried soil was added and 2.5 ml of crude oil was mixed evenly. These steps were repeated for 4 times until the approximate concentration of crude oil in the soil was 10 ml/100 g of soil. The crude oil contaminated soil was placed in fume hood for 7-10 days to oil adsorption in the soil.

In situ removal of crude oil from contaminated soil

Growth medium of 50 ml sterile Zobell broth and 10 g of contaminated soil (containing 1 g of oil) was taken in 500 ml sterile flask and inoculated with 80 × 104 (20 µl) of cells and incubated for 48 hours in orbital shaker at 37°C. After 48 hours, flask culture was centrifuged and the supernatant was subjected to oil removal analysis.

Oil contaminated soil washing with cell free supernatant and partially purified bio surfactant

The soil washing study was conducted to observe crude oil removal with the cell free supernatant and partially purified biosurfactant. 10 g of soil mixed with 50 ml of cell free supernatant and partially purified biosurfactant individually. Every 30 mins the aqueous part was removed and subjected to analysis.

Comparison of synthetic surfactants and bio surfactant for the removal of crude oil

In order to determine the removal of crude oil, the surfactants such as SDS, Tween 80, triton-X 100, Cetyl Trimethyl Ammonium Bromide (CTAB), phospholipid and biosurfactant were isolated from marine Bacillus sp and was taken for the comparison studies for the removal of crude oil. Three different concentrations (1, 5 and 10%) of surfactants and biosurfactants were carried out for the oil removal study.

Shaking and static condition

Different concentration of surfactant (both synthetic and bio-) were taken in the 500 ml of flask containing 1 gm contaminated soil and incubated individually under shaking (200 rpm) condition. Every 30 mins, they were centrifuged, subsequently aqueous and soil were separated which was mixed with hexane (1:1) and it was subjected to separation technique for the oil removal.

Different concentrations (%) of surfactant (both synthetic and bio-) were taken in the 500 ml of flask which contains 1 gm contaminated soil and was incubated under static (200 rpm) condition individually.

The flask was centrifuged at 200 rpm for 6, 12, 24 and 48 hours and separated aqueous and soil, the residual oil was (extracted with hexane) subjected for oil removal analysis.

Determination of oil removal study

The oil removal was determined by solvent extraction using n-Hexane. Equal amount of n-Hexane was added to the separated aqueous samples and then shaken laterally. The hexane portion was separated and repeated thrice. The entire extract sample was collected and the absorbance of the extract samples were measured at 420 nm using a stock solution of n-hexane/crude oil mixture, which showed the highest absorbance occurred at 420 nm.

The concentration of crude oil at this absorbance was determined from the function obtained from the calibration curve of the stock solution n-hexane/crude oil at 200°C. The percentage of crude oil removal from the water and soil was calculated using the equation.

\[
\text{Crude oil removal} \% = \left( \frac{O_i - O_r}{O_i} \right) \times 100\%
\]

\(O_i\) is the initial oil in the soil (g)/water before washing and \(O_r\) is the oil remaining in the soil (g)/water after washing.

Column chromatography technique

Four different types of soil (sand, beach, clay and clay loamy) were packed in the column (of 33 cm height and 2.5 cm width) individually. Different concentrations (%) of surfactant (both synthetic and bio-) were taken for the oil removal. Surfactant solution was poured and the fraction was subjected to oil removal analysis.

Determination of oil removal by UV-Visible spectrophotometer

UV-Visible spectral analysis of bio surfactants using Shimadzu UV-2450 UV-Visible spectrophotometer with the spectral range of 200-800 nm.

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Determination of oil removal by FT-IR spectral analysis

Biosurfactants samples mixed with KBr (Sigma, USA) and the pellet obtained after hydrolytic pressure was analyzed using Spectrum one (Perkin-Elmer Co., USA model). All measurement consists of 500 scans and plain KBr pellet was used as the background reference.

RESULTS AND DISCUSSION

Before Bacillus sp., was isolated from marine sediment and it was sub cultured in Zobell marine broth and stored in 30% glycerol stock for future use. Morphology of Bacillus sp. has rough, dry, colonies. Gram positive rode of length >160 µm and breadth of 400 nm observed under scanning electron microscope (Figures 1a and 1b).

![Figure 1](image.jpg) Growth of marine Bacillus sp., on Zobell agar medium & SEM image of Marine Bacillus sp. Note: a) Growth of Marine Bacillus sp. (MTCC 5514) on Zobell agar medium; b) SEM image of Marine Bacillus sp.
Several researchers have isolated biosurfactant produced strains belonged to 8 different genera, i.e. Bacillus, Rhodococcus, Halomonas, Alcanivorax, Exiguobacterium, Halomonas, Pseudomonas and Streptomyces. Among them, three were recently established, i.e., Alcanivorax, Exiguobacterium and Halomonas. The genus Alcanivorax is particularly young and was established in 1998 [12], with only 11 species identified so far. All of them were isolated from marine environments and were found to be important alkane degrading bacteria [13], India [14], Iran [15] and Germany [16], most of which are from inland reservoirs. The reported biosurfactant producers in reservoirs included Bacillus licheniformis strains [17,16], Bacillus cereus strain [17], Bacillus subtilis strains [18,19]. Those biosurfactant producers identified from oil field samples were confirmed as the genera of Bacillus sp.

### Instrumental analysis of extracted sample

Crude oil and crude oil transformed samples were subjected to UV-visible spectroscopy, FT-IR and GC analysis. Bacillus sp was isolated from marine sediment and it was subcultured in Zobell marine broth and stored in 30% glycerol stock for further use. Morphology of Bacillus sp has rough, dry, filamentous with pink pigmented colonies. Gram positive rods of length >150 µm and breadth of >400 nm observed under scanning electron microscope. The isolate follows typical growth pattern and the stationary phase achieved only after 180 minutes of incubation.

During growth phase there was increase in the media pH and the final pH was observed as >8.5 ± 0.4. The qualitative activity was analyzed using the drop collapse test which showed that the oil was collapsed immediately when placed on 10 µl of surfactant solution. Surface activity measurements made during the growth of the isolate exhibit surfactant activity after 24 hours of incubation and the maximum surfactant activity observed after 48 hours and maintained till 72 hours. The maximum surface activity of the marine isolate was measured 28 ± 5 mN/m. CMC is a direct indication of the surface activity of a surfactant product. The lower the CMC, the smaller quantity of the surfactant is needed to reach the micellar stage, thus the greater the associated surface activity [20].

Figure 2 illustrated that the bioreactor in single batch which contains 1600 ml of Zobell marine broth which was prepared under aseptic condition. 2 ml of the inoculum was added to the sterile media. The growth was observed after 48 hours interval with reference to the yield of the biosurfactant. The ethanol precipitation procedure was followed which exhibits 35-40 g/L (wt weight) of the surfactant and after lyophilization and drying 46 g/l of biosurfactant was obtained. The surface activity, which result in a broad spectrum of potential applications in oil contaminated control (Figure 2) [21].

Figure 3 illustrated surface tension plot obtained from log concentration of biosurfactants versus surface tension measured in mN/m. The critical micellar concentration was obtained at 2.301 log concentration of biosurfactants (Figure 3).

With respect to the optimization of cell concentration, different concentrations of cells were inoculated in 50 ml of broth with 10 ml of oil were given and incubated individually for 48 hours. In which the 20 µl (80 x 10^9 cells) is the minimum concentration required for the removal/ solubilization of crude oil in aqueous medium (Figure 4).

With respect to the optimization of oil concentration, different concentrations of oil were inoculated in 50 ml of broth with 20 µl of cell concentration were given and incubated individually for 48 hours. Maximum percentage of removal/solubilization of crude oil was observed in 1 ml aqueous medium (Figure 5).

After the incubation period of 48 hours, the crude oil was solubilized and transformed into thread like structures. Figure 6 illustrated the removal of crude oil contaminated soil within 48 hours of incubation. Similar to the aqueous medium, removal of crude oil in contaminated soil is above 90% within 48 hours of incubation.

With respect to comparison of surfactants at different concentration and time interval under shaking condition, higher incubation, concentration gives appreciable removal oil. Bio surfactants are equally good when comparing to the synthetic surfactant under shaking conditions. More than 80% of crude oil was removed with Phospholipid, SDS and Tween. More than 70% of crude oil was removed with Triton-X100 and CTAB. Bio surfactant obtained from the saline microbial source removed more than 90% of the crude oil (Figure 6).

With respect to static condition, more than 90% of the crude oil which was mixed with biosurfactant, phospholipid, SDS and Triton X 100 was removed. Less than 30% removal was observed with Tween and CTAB. At higher concentration and higher incubation time between 80 can remove more than 80% of the crude oil but CTAB was able to remove only less than 50%.

With respect to the column technique, there has been no removal of surfactant when the concentration is around 1% and at higher concentration the hexane extract became colloid, the reason why it is difficult to estimate the removal of crude oil. At 5% appreciable removal of biosurfactant was observed when compared to others. In beach soil, 56, 59 and 68% of crude oil was removed with Tween-80, SDS and biosurfactant respectively. In sandy soil, more than 55% of oil was removed with Triton, SDS and biosurfactant. No much removal in clay and clay loamy soil. In clay and clay loamy soil, more than 30% of oil was removed with triton-x100 and biosurfactant from marine sources. Figure 7 illustrated that biosurfactant treated and untreated soil samples. When surfactants present in a dynamic equilibrium between the continuous phase (water) and the disperse (oil) phase at interface, the surfactants with relatively high affinities towards aqueous phase can desorb from a droplet and diffuse through the aqueous phase, especially in marine oil spill scenarios when water is typically more abundant [22,23]. This may leave the droplet interfaces relatively vulnerable for coalescence [24]. Therefore, surfactants with higher affinity towards the oil phase may have a slight upper-hand in this regard, such as lecithin (Figure 7) [22].

### Instrumental analysis of crude oil and bio transformed sample

#### Spectral analysis UV-Visible

The electrostatic charge plays a primary role in sustaining cell activities and behaviors through influencing the overall cell polarity and maintaining the degree of surface hydrophilicity [25]. The electrical potential of the interfacial region between the bacterial surface and the aqueous environment has been widely used to assess the net cell surface charge [25]. Dissipation of potential of Bacillus sp biomasses were assessed and the results are presented in Figure 8 illustrated that the crude oil and crude oil treated with biosurfactant sample which is extracted with hexane and subjected to UV-Visible spectrum. UV-Visible spectrum analyses showed presence of protein compounds (a small hump in the region of 280-290 nm). There is change in the crude oil peaks and some peaks also shifted (Figure 8).
Figure 3) Biosurfactant activity and yield of marine Bacillus sp. Notes: (△) pH, (○) Surface tension, (□) Yield.

Figure 4) Surface tension plot

Figure 5) Optimization of cell concentration

Figure 6) Optimization of oil concentration
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FTIR spectral analysis

They are lately hydrated and subsequently convert to the forms of carbonates and phosphates [26]. The alkane C-H bond stretch (2700-3000 cm⁻¹) and carboxylic and/or hydroxyl groups (3200-3600 cm⁻¹) were identified from the FTIR analysis. The presence of Al and Fe as oxygen functional groups within Al-OH and Fe-OH (800-900 cm⁻¹) can be confirmed from Putra et al., [27]. The FTIR spectrum of samples after the incubation process was also presented in FTIR analysis of the crude oil showed peaks at 3000-2850, 1680-1600, 1400-1000, 1350-1000, 1000-650 cm⁻¹ corresponding to C-H, C=C, C≡C, C=O, C-H, C-N, =C-H respectively. Figure 8 illustrated that the hexane extracted sample from the crude oil contaminated soil showed peaks at 3650-3600, 3500-3100, 2270-1940, 1690-1640, 1350-1140, 785-540 cm⁻¹ corresponding to O-H, N-H, C=O, C≡C-C≡C, C-H, N-H, S=O, C-H respectively [28].

CONCLUSION

This study helped to explain the biologically derived surface-active agents on removal of oil from oil spilled soil. In this study, compared with using crude oil removal rate was significantly increased using biosurfactant based washing solution. The mechanism of biosurfactant enhanced crude oil removal is closely related with its concentration. When the concentration of biosurfactant solution was below its Critical Micelle Concentration (CMC) value, the mechanism of biosurfactant enhanced crude oil removal. The lowered interfacial tension thus led to an increased contact angle and reduced capillary force holding the crude oil and soil particles and consequently enhanced the mobility of crude oil. When the concentration of biosurfactant is above its CMC value, the formation of biosurfactant micelle can greatly increase the solubilization process, and help to solubilize the residue oil compounds left in the soil system and enhance the removal of organic contaminants.

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